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Synthesis and Reactivity of Backfluorinated NHC Carbene Complexes

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Introduction

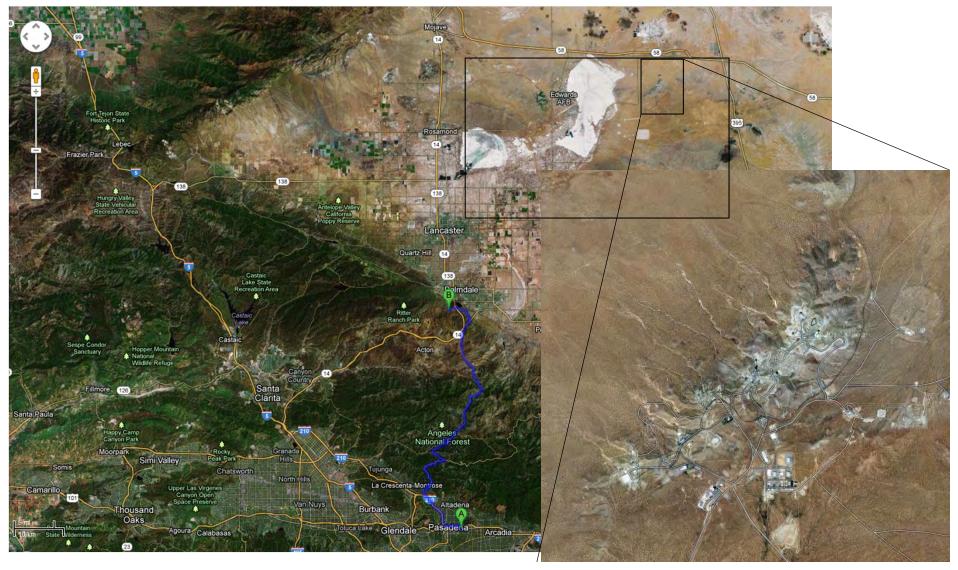


- Introduction
- Proposed Research Overview
- Synthesis of Backfluorinated NHC Carbene Precursors
 - Synthesis of Ir complexes
 - IR study of Ir dicarbonyl complexes
 - Synthesis of Ru complexes
 - Reactivity of Ru complexes
- Conclusions and Future Directions



Edwards AFB – AFRL Rocket Propulsion Research







Palmdale to AFRL: 53 miles Located in the Middle of Nowhere







Edwards AFB



History

- Originally known as Muroc Army Air Corps Base
- Test flights of the YB-42 (first American Jet) in the early '40s
- Location where Chuck Yeager broke the Sound Barrier in the Bell X-1 (Original craft at the Smithsonian)
- X-15 sub-orbital flights in the '60s (Armstrong)

AFRL

- Tenant of Edwards AFB since late '50s
- Full scale testing of the Atlas rockets (Gemini missions)
- Initial testing of the F-1 engine (Apollo missions) performed on site in the mid '60s
- Large scale testing of solid rocket motors (Titan IV)
- "Iranian nuclear facility" destroyed by the Transformers in "Transformers: The Dark side of the Moon"



Edwards AFB – AFRL Projects



- The Air Force has an interest in NHC carbene precursors for a variety applications
 - lonic liquid propellants and additives
 - Ligands for Supercritical Chemical Fluid Deposition (SCFD)
- The Air Force also has an interest in fluorinated NHC carbenes
 - perfluoroalkyl chains generally known to improve solubility of systems in supercritical fluids
 - "Backfluorinated" NHC carbenes to improve solubility in supercritical fluids and perhaps maintain ligand stability
 - Surprisingly, the backfluorinated systems have never been reported



Potential for Collaboration Backfluorinated NHC-Ru Catalysts



- It was an easy leap to see that this technology should be applicable to other research areas
 - Hydrogenation in fluorinated solvents
 - Organic transformations in fluorinated solvents
 - Olefin metathesis in fluorinated solvents
 - Perhaps improved olefin metathesis of fluorinated olefins

Questions

- How does the addition of a perfluoroalkyl chain to the back of an imidazolinylidene ligand affect its electronic properties?
- How does the addition of a perfluoroalkyl chain to the back of an imidazolinylidene ligand affect the catalytic properties of the ruthenium alkylidene complexes?
- Does the addition of perfluoralkyl chains improve the solubility of Ru alkylidene complexes in fluorinated solvents?



Fluoroalkene Metathesis



- Fluoroalkene metathesis first reported by Blechert in 2001
- 1st generation systems not effective
- 2nd generation systems are effective
 - 2nd Generation Grubbs-Hoveyda most effective



Fluoroalkene Metathesis



- Blechert (2001) found that cross metathesis of perfluorobutyl(ethene) and various allylics required 10 mol% of catalyst with generally good yields (7-95+%)
- Kotora (2010) found that the cross metathesis of perfluorohexyl(propene) and various allylics was effective for several systems but the yields varied considerably (11-70% yield)
- Also, when the perfluoroalkyl chain is longer (perfluorooctyl-, perfluorodecyl-), personal research suggests that the catalyst has solubility issues

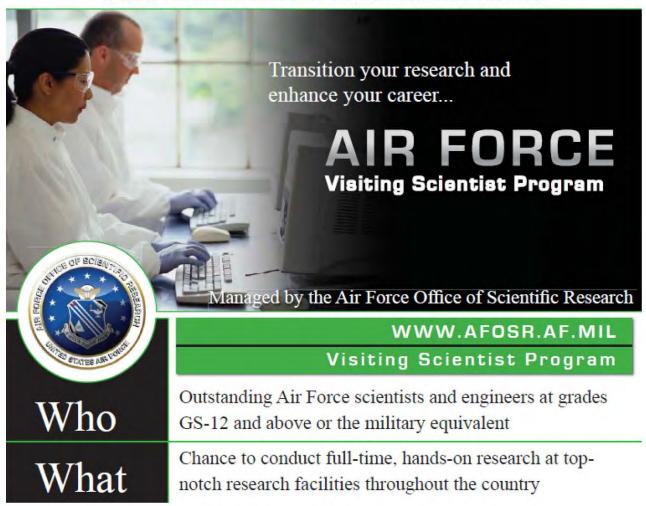


AFOSR Visiting Scientist Program 6 Month Sabbatical to Research Institutions



AFOSR-TODAY'S BREAKTHROUGH SCIENCE FOR TOMORROW'S AIR FORCE

SPECIAL PROGRAMS: AIR FORCE VISITING SCIENTIST PROGRAM





Proposed Research



- Electronic effects will be investigated
 - Effect of perfluoroalkyl length and methylene "buffer" length
- Olefin metathesis activity will also be investigated



Synthetic Targets Imidazolinidene Complexes



$$F_3C - \left(CF_2\right)_3 \qquad F_3C - \left(CF_2\right)_5 \qquad F_3C - \left(CF_2\right)_7 \qquad F_3C - \left(CF_2\right)_9 \qquad \qquad \\ Mes - N - Mes \qquad Mes - Mes \qquad Mes - N - Mes \qquad Mes - N - Mes \qquad Mes - N - Me$$

Perfluoroalkyl chain length and "Buffer" length will be investigated



Preparation of Perfluoroalkyl Grignard Reagents



$$FCF_{2}CH_{2}CH_{2}CI \qquad \longrightarrow \qquad FCF_{2}CH_{2}CH_{2}MgCI$$

$$reflux \qquad \qquad FCF_{2}CH_{2}CH_{2}MgCI$$

$$F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}I \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}MgI$$

$$n=3,4 \qquad \qquad Mg, thf$$

$$F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}Br \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}MgBr$$

$$n=2,5 \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}Br$$

$$n=2,5 \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}Br$$

$$n=4 \qquad \longrightarrow \qquad Mg, thf$$

$$reflux \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}Br$$

$$n=4 \qquad \longrightarrow \qquad Mg, thf$$

$$reflux \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}CH_{2}Br$$

$$n=4 \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}CH_{2}Br$$

$$reflux \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}CH_{2}Br$$

$$reflux \qquad \longrightarrow \qquad F(CF_{2}CF_{2})_{n}CH_{2}CH_{2}CH_{2}CH_{2}Br$$

n=4

n=4



Synthetic Method #1 Grignard Addition/Reduction/Cyclization



$$Mes = Mesityl$$

$$R' = F_3C + CF_2 \choose n$$

$$Mes = Mesityl$$

$$R' = F_3C + CF_2 \choose n$$

$$Mes = Mesityl$$

$$R' = F_3C + CF_2 \choose n$$

$$Mes = Mesityl$$

$$N = Mes$$

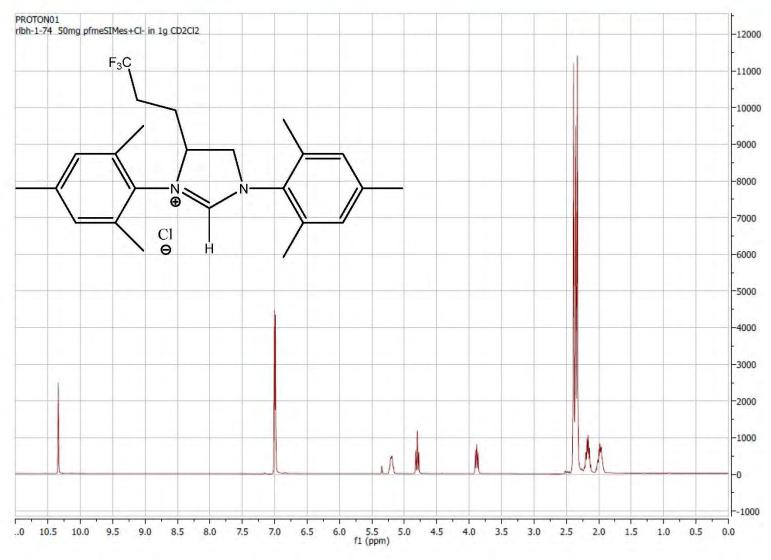
$$N$$

- Very general reaction pathway: suitable for all Grignard reagents
- Attempts at dialkylation unsuccessful
- Slight excess of Grignard reagent ensures complete monoalkylation



¹H Spectrum of pfme-SIMes+CI-

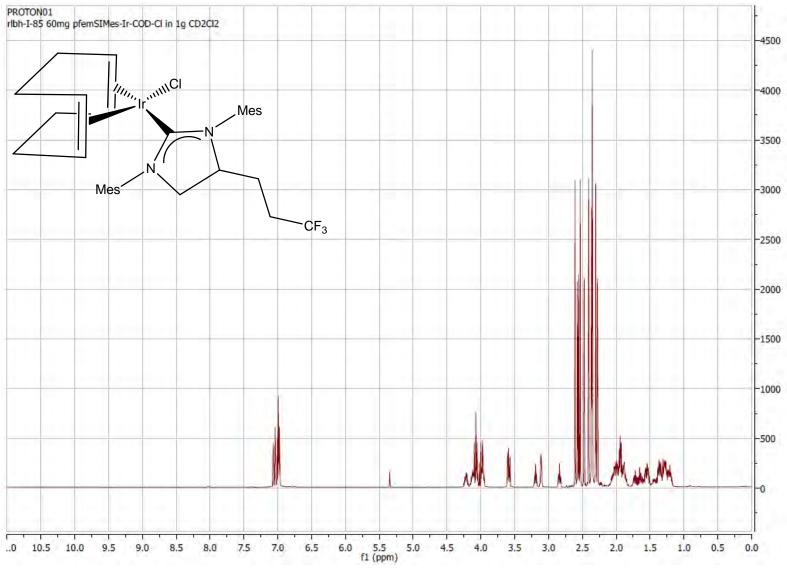






¹H Spectrum of pfme-SIMes-IrCOD-CI

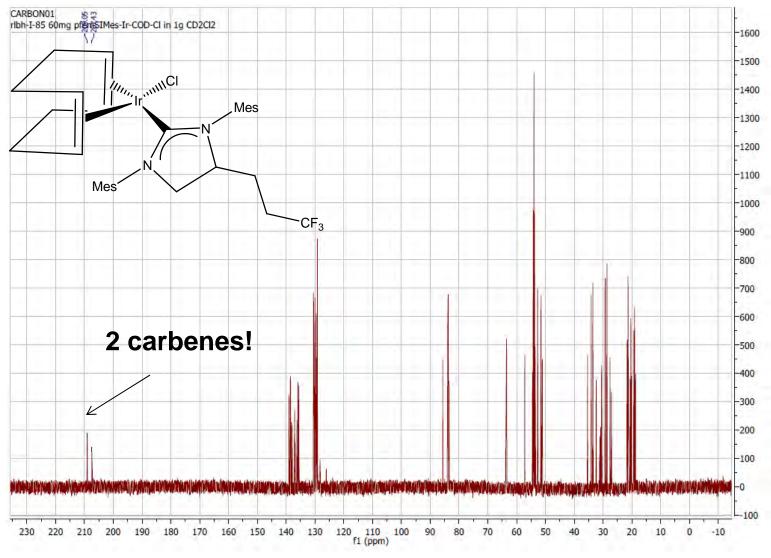






¹³C Spectrum of pfme-SIMes-IrCOD-CI

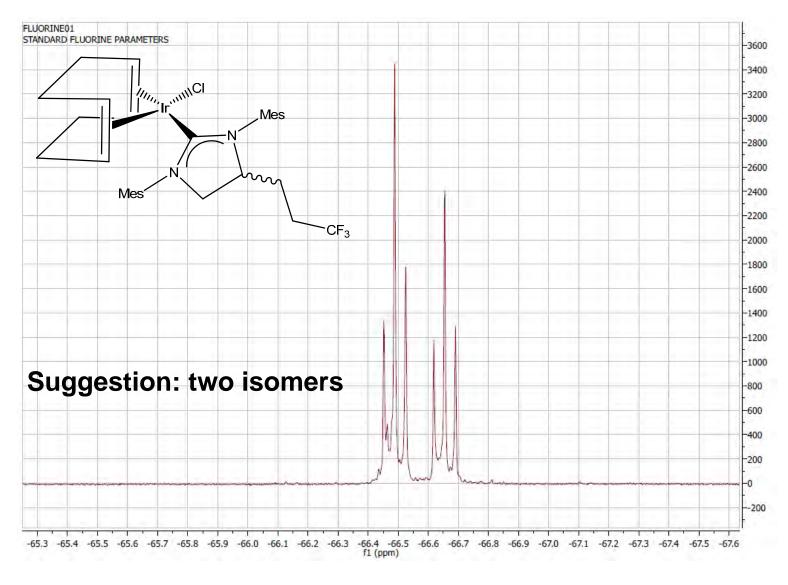






¹⁹F Spectrum of pfme-SIMes-IrCOD-CI

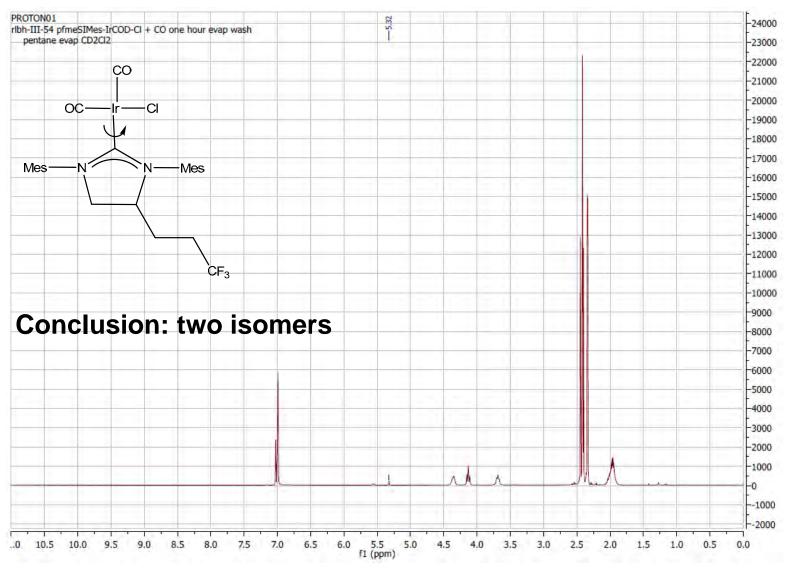






¹H Spectrum of pfme-SIMes-Ir(CO)₂-CI







Potential Isomers in Solution



Ligand rotation observed upon heating in toluene-d₈



Backfluorinated NHC Complexes with (CO)₂-Ir-Cl in CD₂Cl₂(cm⁻¹)



Compound	Functional Group	νCO 1 st CO	νCΟ 2 nd CO	νCO Avg.
SIMes	Н	2068.0	1981.2	2024.6
pfme-SIMes	CF ₃ CH ₂ CH ₂	2068.8	1981.9	2025.4
pfbe-SIMes	F(CF ₂) ₄ CH ₂ CH ₂	2068.8	1981.9	2025.4
pfhe-SIMes	F(CF ₂) ₆ CH ₂ CH ₂	2069.1	1982.0	2025.6
pfoe-SIMes	F(CF ₂) ₈ CH ₂ CH ₂	2069.1	1982.0	2025.6
pfde-SIMes	F(CF ₂) ₁₀ CH ₂ CH ₂	2069.1	1982.1	2025.6
pfop-SIMes	F(CF ₂) ₈ CH ₂ CH ₂ CH ₂	2068.5	1981.2	2024.8

When buffer length is CH₂-CH₂, the perfluoroalkyl group appears to have a small electronic effect



Synthesis of Ru Complexes



- Green solid (same color as 2nd generation catalyst)
- Complex demonstrates metathesis activity
- Perfluorodecylethyl, perfluorooctylethyl and perfluorooctylpropyl complexes also synthesized



Partition Study





- Partition study with pfheSIMes 2nd Generation Grubbs-Hoveyda catalyst with toluene (upper phase) and perfluoromethyl cyclohexane (lower phase)
 - Insufficient backfluorination to improve solubility in perfluoromethylcyclohexane



Backfluorinated NHC Carbene complexes



Conclusions

- A series of backfluorinated imidazolinium NHC carbene complexes were synthesized
- Each backfluorinated NHC carbene-COD-iridium chloride complex is a mixture of two observable isomers due to the lack of rotation of the NHC carbene ligand
- The backfluorinated NHC carbene-dicarbonyliridium chloride complex is single compound due to rotation of the NHC carbene ligand
- An IR study of the backfluorinated NHC carbenedicarbonyl-iridium chloride complexes determined that the addition of a perfluoroalkyl chain slightly changes the electronics of the molecule.



Backfluorinated NHC Carbene complexes



Conclusions – Cont'd

- The IR study of the backfluorinated NHC carbenedicarbonyl-iridium chloride complexes also suggest that the electronic effects of the perfluoroalkyl group is independent of perfluoroalkyl chain length
- A series of the longer chain backfluorinated NHC Ruthenium alkylidene complexes were synthesized
- These complexes demonstrated metathesis activity similar to the nonflourinated alkylidene complex



Backfluorinated NHC Carbene Complexes



Future Work

- Look for methods to increase the amount of backfluorination of imidazolinium complexes
- Investigate the electronic effects of side fluorination of aromatic rings of imidazolinium complexes
- Investigate the use of the technology for other applications in order to improve sustainability



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